# Sodium Hydroxide Catalyzed *N*-Alkylation of (Hetero) Aromatic Primary Amines and N$_1$,C$_5$-Dialkylation of 4-Phenyl-2-aminothiazoles with Benzyl Alcohols

Ramachandra Reddy Donthiri, Venkatanarayana Pappula, Darapaneni Chandra Mohan, Hiren H. Gaywala,[‡] and Subbarayappa Adimurthy*

Academy of Scientific & Innovative Research, CSIR−Central Salt & Marine Chemicals Research Institute, G.B. Marg, Bhavnagar-364 002, Gujarat, India

S *Supporting Information*

ABSTRACT: In the presence of a catalytic amount of NaOH, the selective *N*-alkylation of various heteroaromatic primary amines is reported. With 1 equiv of NaOH, N$_1$,C$_5$-dialkylation of 4-phenyl-2-aminothiazoles has been investigated. Reaction of in situ generated aldehyde with amine yields the *N*-alkylated and N$_1$,C$_5$-dialkylated products through hydride ion transformation from alcohol.
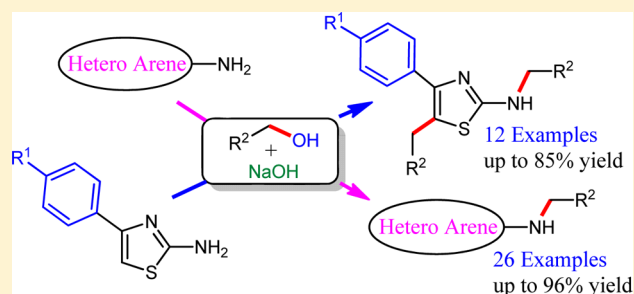
The C−N bond-forming reactions have attracted much attention over the last two decades and have found its applications in all areas of organic chemistry, ranging from the laboratory bench to the synthesis of pharmaceutical fine chemicals and the production of bulk chemicals.[1−4] Among the C−N bond forming reactions, the *N*-alkylation of primary amines is important because of the tremendous utility of these amines in industry for the synthesis of dyes, additives, agrochemicals, functional materials, and pharmacophores for the synthesis of bioactive molecules.[5−8] Arguably, the direct *N*-alkylation of amines with alcohols is most desirable because of the ready and inexpensive availability of wide range of alcohols and theoretically produces only water as a byproduct. In this regard, many groups have achieved *N*-alkylation of amines with alcohols using transition-metal/metal complexes as catalysts through borrowing hydrogen methodology.[9] In addition to these recent advancements, a major drawback of these methods is the requirement of transition-metal catalysts to promote this transformation efficiently. Thus, transition-metal separation from organic products is of particular importance for the synthesis of pharmaceutical fine chemicals because their residual toxicity in the target compounds is a central issue to consider; hence, the development of more improved synthetic routes without transition metals is desirable. Although transition-metal-free *N*-alkylation of amines with alcohols has been known, the reactions demand harsh conditions such as high temperature (250−300 °C), high pressure, very long reaction time (10 days), strong basic conditions, and large excess of alcohols or amines; consequently, low yields and poor selectivities have been reported.[10−16] Very recently, Xu and co-workers reported the transition-metal-free *N*-alkylation of sulfonamides and aromatic amines in the presence of base

(10−100 mol %) and external addition of catalytic amount of aldehyde.[17]

Recently, we have reported the transition-metal-free imine synthesis through activation of alcohol by base in presence of amines.[18,19] In continuation of our efforts to explore C−N bond-forming reactions,[20−24] we sought to investigate the direct *N*-alkylation of heteroaromatic primary amines with alcohols using NaOH as catalyst (20 mol %) under transition-metal-free conditions.

2-Aminobenzothiazoles and thiazoles are versatile synthetic intermediates and are widely used in the synthesis of various biologically active compounds.[25−30] In particular, *N*-alkylated thiazoles and benzothiazoles exhibit different pharmacological and physiological activities.[31,32] As a result, much attention has been paid to the synthesis of these compounds.[33−39] By considering the importance of 2-aminobenzothiazole derivatives, we investigated herein the *N*-alkylation of 2-amino-benzothiazoles, and the method has been extended to other heteroaromatic amines including pyridine, pyrimidine, and pyrazines with various alcohols. Under these conditions, we found the N$_1$,C$_5$-dialkylation of 4-phenyl-2-aminothiazoles, and there were no reports found for this type of transformation.

At the start of our studies, we investigated the *N*-alkylation of 2-aminobenzothiazole **1a** with 4-chlorobenzyl alcohol **2a** as alkylating agent, and results are illustrated in Table 1. In the presence of NaOH in toluene at room temperature no product **3a** was formed (Table 1, entry 1). When the reaction was performed at 100 °C, the *N*-alkylated product was observed with 80% GC yield (Table 1, entry 2). Conversion reached

Table 1. Optimization of Reaction Conditions for 3a[a]
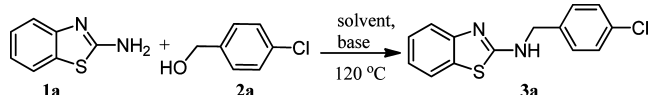
entry	base	solvent	yield[b] (%)
1[c]	NaOH	toluene	
2[d]	NaOH	toluene	80
3	NaOH	toluene	100 (93)
4	NaOH	DMF	
5	NaOH	DMSO	
6	NaOH	H₂O	
7	NaOH	glycerol	
8	NaOH	chloro benzene	70
9	KOH	toluene	85
10	Na₂CO₃	toluene	
11	K₂CO₃	toluene	30
12	NaOᵗBu	toluene	87
13	KOᵗBu	toluene	95
14[e]	**NaOH**	**toluene**	**100 (93)**
15		toluene	
16	Na metal	toluene	100
17	NaH	toluene	100
18[f]	NaOH	toluene	100 (90)
19[e,g]	NaOH	toluene	2
20[e,f,g]	NaOH	toluene	1

[a]Reaction conditions unless otherwise stated: **1a** (1.0 mmol), **2a** (1.2 mmol), base (1.0 mmol), solvent (1.0 mL), 120 °C, 15 h, under atmospheric air. [b]GC yields. [c]Room temperature. [d]100 °C. [e]NaOH (20 mol %). [f]Nitrogen atmosphere. [g]Without **1a** yield refers to corresponding aldehyde (see Figures S2−S6 in Supporting Information for entries 19 and 20). Numbers in parentheses are isolated yield.

100% (93% isolated) when the reaction temperature was increased to 120 °C (Table 1, entry 3). However, the reaction does not proceed efficiently in other solvent systems (Table 1, entries 4−8). Alternatively, screening of different bases (KOH, Na₂CO₃, K₂CO₃, KOᵗBu, and NaOᵗBu) was performed (Table 1, entries 9−13). After extensive screening of various bases and solvents, NaOH showed good activity and selectivity in toluene at 120 °C, and a similar yield (93%) was observed even with 20 mol % of NaOH (Table 1, entry 14). The reaction was completely retarded in the absence of NaOH (Table 1 entry 15). To rule out the catalytic activity of trace impurities of transition metals in commercially available NaOH, two experiments were performed using pure Na metal and NaH (Table 1, entries 16 and 17). These species also provided similar yield and selectivity of the product for the present transformation. These facts support the theory that the role of catalytic activity of trace metal impurities from the commercial NaOH is less probable. To examine the role of oxygen, the reaction was performed under nitrogen atmosphere; it provided 90% isolated yield of **3a** (Table 1, entry 18). Further, **2a** was subjected to oxidation in the absence of amine under open and nitrogen atmosphere, and only 2% and 1% yield of aldehyde was observed, respectively (Table 1, entries 19 and 20). Oxidation under nitrogen atmosphere may be due to the presence of trace amount of oxygen from the commercial cylinder. These reactions clearly indicate that the in situ generated catalytic aldehyde is responsible for the current transformation.

The scope of the reaction was then investigated with 2-aminobenzothiazoles and benzyl alcohols as alkylating agents

(Table 2). Under these optimized conditions, various substituted benzyl alcohols reacted with 2-aminobenzothiazole

Table 2. N-Alkylation of Various Heteroaromatic Primary Amines with Benzyl Alcohols[a]
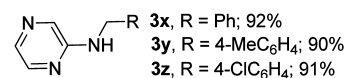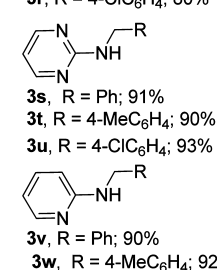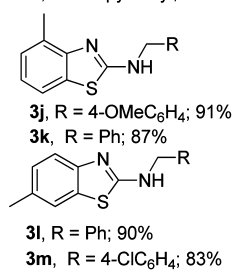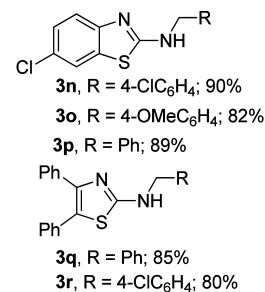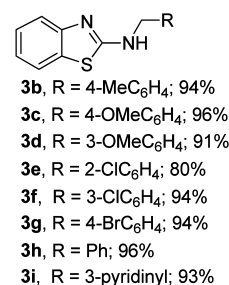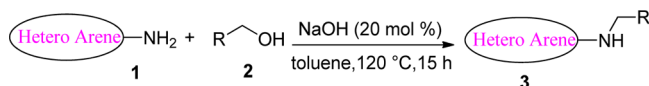
[a]Reaction conditions unless otherwise stated: amine (1 mmol), alcohol (1.2 mmol), NaOH (20 mol %), toluene (1 mL), 120 °C, 15 h; isolated yield.

to yield the corresponding N-alkylated products in good yield. Benzyl alcohols having electron-donating (methyl or methoxy) groups transformed into N-alkylated 2-aminobenzothiazoles in high yield (**3b−d**). Electronic effects associated with electron-withdrawing substituents (Cl, Br) on the phenyl ring (either ortho/meta/para) do not affect the efficiency of the N-alkylation of 2-aminobenzothiazole (**3e−g**).

Unsubstituted benzylic alcohol and heterobenzylic alcohol (pyridin-3-ylmethanol) also provided the desired products in excellent yield (**3h, 3i**). The N-alkylation of different substituted 2-aminobenzothiazoles was then studied. N-Alkylation of 2-aminobenzothiazoles bearing electron-releasing (Me) and electron-withdrawing (halogen) groups afforded the corresponding N-alkylated products in 82−91% yield (**3j−p**). Remarkably 4,5-diphenyl-2-aminothiazole also gave the required products in good yields (**3q,r**). Notably, the products **3a−r** were purified without column chromatography (see the typical procedure). Further, we have extended the present transformation to other heteroaromatic amines such as 2-aminopyrimidine, 2-aminopyridine, and aminopyrazine. These heterocyclic amines also afford good to excellent yields of N-alkylated products (**3s−z**) irrespective of alcohols under the optimized conditions. No desired products were observed with other alcohols such as cyclohexanol, n-heptanol, 3-methylbut-2-en-1-ol, and cinnamyl alcohol.
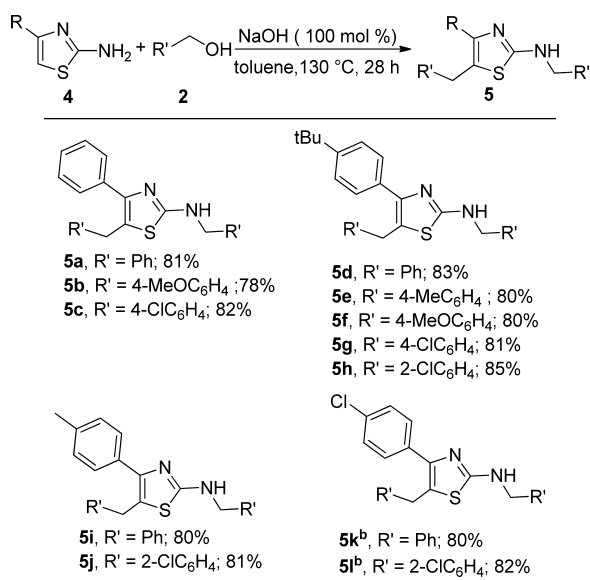
Having developed a synthetic method for N-alkylation of heteroamines, we finally explored the current concise trans-

formation for dialkylation of **4a** (Scheme 1). The optimized conditions of Table 2 were applied to **4a** with benzyl alcohol as

Scheme 1
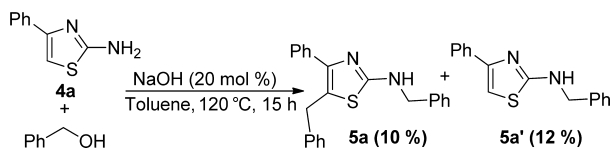
alkylating agent, and a mixture of di- and monoalkylated products **5a** and **5a′** was obtained. Under these conditions, the formation of dialkylated product **5a** was quite interesting, and no such reports exist in the literature. Hence, we became interested in investigating the conditions for selective formation of a single dialkylated product **5a**. The identified conditions for this transformation were 3 equiv of benzyl alcohol, 1 equiv of amine **4a** at 130 °C in toluene, and 28 h of reaction time. These revised optimum conditions were applied to explore the synthesis of N_1,C_5-dialkyl-4-phenylthiazol-2-amines (Table 3).

Table 3. N_1, C_5-Dialkylation of 4-Phenyl 2-aminothiazoles with Various Alcohols[a]

5a, R′ = Ph; 81%
5b, R′ = 4-MeOC₆H₄ ;78%
5c, R′ = 4-ClC₆H₄; 82%

5d, R′ = Ph; 83%
5e, R′ = 4-MeC₆H₄ ; 80%
5f, R′ = 4-MeOC₆H₄; 80%
5g, R′ = 4-ClC₆H₄; 81%
5h, R′ = 2-ClC₆H₄; 85%

5i, R′ = Ph; 80%
5j, R′ = 2-ClC₆H₄; 81%

5k[b], R′ = Ph; 80%
**5l[b]**, R′ = 2-ClC₆H₄; 82%

[a]Reaction conditions unless otherwise stated: amine **4** (1 mmol), alcohol **2** (3 mmol), NaOH (1.0 mmol), toluene (1 mL), 130 °C, 28 h. Isolated yield. [b]Reaction time 36 h.

It was found that N_1,C_5-dialkylation of **4a** proceeded smoothly with benzylic alcohols having electron-releasing and -withdrawing substituents (**2a−c**) to give the corresponding products (**5a−c**) in good yields. Similarly, the presence of electron-releasing/withdrawing groups on 4-phenylthiazol-2-amines afforded the corresponding dialkylated products (**5d−l**) in good yields with irrespective of benzylic alcohols. Under these conditions, dialkylation of unsubstituted 2-aminothiazole was not observed as it undergoes charring. The formation of **5h** was further confirmed by single-crystal XRD (see Figure S1, Supporting Information).

The formation of dialkylated product may be due to the electron-rich nature of five-membered thiazole compared to six-membered pyridine, pyrimidine, and pyrazine systems. The substitution of phenyl ring at C_4 may further increase the

electron density on C_5 of **4a**. To further confirm the current transformation, we performed the reaction starting from imine (Scheme 2), under the same reaction conditions, and 70% of

Scheme 2. Mechanistic Experiment
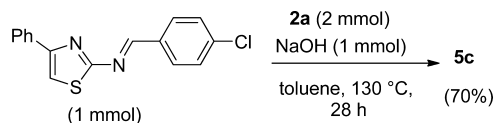
dialkylated product **5c** was isolated along with 68% of 4-chlorobenzoic acid after workup. Based on the above experimental evidence and previous reports,[17−19] we propose a plausible mechanism (Scheme 3). Initially, alcohol was partly

Scheme 3. Possible Mechanism for the N_1, C_5-Dialkylation of 4-Phenyl-2-aminothiazole
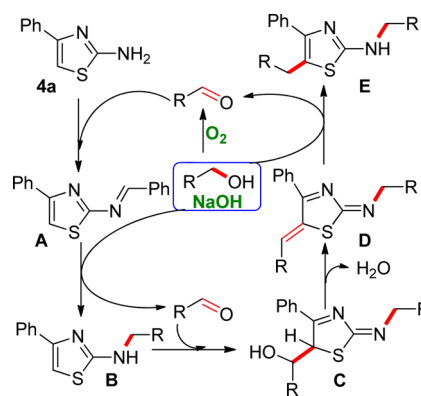
oxidized to aldehyde in presence of NaOH and air as an oxidant.[40,41] The reaction of aldehyde and amine **4a** generate imine intermediate **A**. In the presence of base, dehydrogenation of alcohol with simultaneous hydrogenation of imine lead to aldehyde and N-alkylated product **B**, respectively. Nucleophilic addition of **B** to aldehyde may form imino alcohol **C**, and its dehydration gives another intermediate **D**. Through a hydride ion transfer process, it provides the final product **E**.

In summary, we have developed a novel method for N-alkylation of heteroamines with benzylic alcohols as alkylating agents in the presence of catalytic NaOH under mild conditions. Reaction of 4-phenylthiazol-2-amines with benzyl alcohols gave N_1,C_5-dibenzyl-4-phenylthiazol-2-amines in good yields with 1 equiv of NaOH. Compared to previously known transition-metal catalysts, NaOH is inexpensive, readily available, and convenient to use without any special precautions. The method is applicable for a wide range of benzylic alcohols as alkylating agents.

■ EXPERIMENTAL SECTION

General Methods. All commercially available chemicals and reagents were used without any further purification unless otherwise indicated. ^{1}H and ^{13}C NMR spectra were recorded at 500/200 and 125/50 MHz, respectively. The spectra were recorded in DMSO as solvent. Multiplicity was indicated as follows: s (singlet); d (doublet); t (triplet); m (multiplet); dd (doublet of doublets), etc. Coupling constants (J) are given in Hz. Chemical shifts are reported in δ relative to TMS as an internal standard. The peaks around δ values of ^{1}H NMR (2.5), ^{13}C NMR (39.4) are correspond to deuterated DMSO and δ value (3.3) in ^{1}H NMR is of water. Progress of the reactions was

monitored by thin-layer chromatography (TLC). Silica gel 100−200 mesh size was used for column chromatography.

Typical Procedure for Synthesis of 3a. In a 25 mL round-bottomed flask were placed 150 mg (1.0 mmol) of 2-amino-benzothiazole (**1a**), 171 mg (1.2 mmol) of 4-chlorobenzyl alcohol (**2a**), 8 mg (0.2 mmol) of sodium hydroxide (NaOH), and 1.0 mL of toluene. The reaction flask was heated at 120 °C for 15 h in an oil bath. After completion of the reaction, the flask allowed to attain room temperature, and toluene was removed under reduced pressure. To the residue was added 20 mL of water, the mixture was stirred for 5 min on a magnetic stirrer at room temperature, and the solid was collected through filtration and washed with 10% ethyl acetate in hexane (10 mL) solution to obtain a white solid in 93% yield (255 mg; 0.93 mmol) of **3a**. [Note: The products (**3a−r**) obtained were pure (by NMR analysis) without separation by column chromatography but by simple solvent wash. The products obtained from liquid and solid benzylic alcohols were washed with only hexane and ethylacetate/hexane (1:9), respectively. Products **3s−z** were purified by column chromatography (the eluent for products **3s−w** (9:1) and for the products **3x−z** (7:3) ethylacetate/hexane)].

Characterization Data of All Compounds. *N-(4-Chlorobenzyl)-benzo[d]thiazol-2-amine (3a):*[33] yield 93% (255 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.58(d, J = 6 Hz, 2H), 7.03(t, J = 7.5 Hz, 1H),7.22 (t, J = 7.5 Hz, 1H), 7.38−7.40 (m, 5H), 7.66 (d, J = 7.5 Hz, 1H) 8.54 (t, J = 6 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 46.8, 118.5,121.3, 121.5, 126.0, 128.7, 129.6, 130.8, 131.9, 138.4, 152.7, 166.5.

N-(4-Methylbenzyl)benzo[d]thiazol-2-amine (3b):[33] yield 94% (239 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.27 (s, 3H), 4.54 (d, J = 5 Hz, 2H), 7.02 (t, J = 7.5 Hz, 1H), 7.14 (d, J = 8, 2H), 7.22 (t, J = 8 Hz, 1H), 7.26 (d, J = 8 Hz, 2H), 7.38 (d, J = 8, 1H), 7.65 (d, J = 8 Hz, 1H), 8.47 (t, J = 5, 1H) ; ^{13}C NMR (125 MHz, DMSO) 23.0, 49.3, 120.4, 123.2, 127.8, 129.7, 131.2, 132.7, 138.1, 138.4, 154.8, 168.5.

N-(4-Methoxybenzyl)benzo[d]thiazol-2-amine (3c):[33] yield 96% (259 mg); ^{1}H NMR (500 MHz, DMSO) δ 3.72 (s, 3H), 4.51 (d, J = 5 Hz, 2H), 6.89 (d, J = 7 Hz, 2H), 7.01 (t, J = 7.5 Hz, 1H), 7.21 (t, J = 7.5 Hz, 1H), 7.30 (d, J = 7.5 Hz, 2H), 7.38 (d, J = 8 Hz, 1H), 7.65 (d, J = 8 Hz, 1H), 8.42 (s, 1H); ^{13}C NMR (125 MHz, DMSO) 49.0, 57.3, 116.1, 120.3, 123.2, 127.8, 131.1, 132.7, 133.1, 154.8, 160.7, 168.4.

N-(3-Methoxybenzyl)benzo[d]thiazol-2-amine (3d): yield 91% (246 mg); ^{1}H NMR (500 MHz, DMSO) δ 3.73 (s, 3H), 4.56 (d, J = 5.5 Hz, 2H), 6.82 (d, J = 8 Hz, 1H), 6.94 (d, J = 6.5 Hz, 2H), 7.02 (t, J = 7.5 Hz, 1H), 7.20−7.27 (m, 2H), 7.38 (d, J = 7.5 Hz, 1H), 7.66 (d, J = 7.5 Hz, 1H), 8.49 (t, J = 5.5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 47.1, 54.9, 112.2, 112.9, 118.0, 119.4, 120.8, 125.4, 129.3, 130.3, 140.4, 152.34, 159.2, 166.1; IR (KBr) 672, 751, 796, 978, 1043, 1163, 1264, 1342, 1451, 1566, 1612, 2896, 3464; HRMS calcd for C15H15N2OS 271.0906, found 271.0934.

N-(2-Chlorobenzyl)benzo[d]thiazol-2-amine (3e):[33] yield 80% (219 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.67 (d, J = 5.5 Hz, 2H), 7.03 (t, J = 7.5 Hz, 1H), 7.21 (t, J = 7.5 Hz, 1H), 7.30−7.35 (m, 2H), 7.38 (d, J = 8.0 Hz, 1H), 7.47 (s, 2H), 7.67 (d, J = 7.5 Hz, 1H), 8.53 (s, 1H); ^{13}C NMR (125 MHz, DMSO) 47.3, 120.5, 123.3, 123.4, 127.8, 129.5, 131.2, 131.4, 131.5, 132.8, 134.6, 138.2, 154.6, 168.3.

N-(3-Chlorobenzyl)benzo[d]thiazol-2-amine (3f): yield 94% (257 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.60 (d, J = 6 Hz, 2H), 7.04 (t, J = 7.5 Hz, 1H), 7.23 (t, J = 7 Hz, 1H),7.33 (t, J = 8.5 Hz, 2H), 7.37−7.40 (m, 2H), 7.42 (s, 1H), 7.67 (d, J = 8.0 Hz, 1H); 8.56 (t, J = 6 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 46.9, 118.6, 121.4, 126.0, 126.4, 127.4, 127.5, 130.7, 130.8, 133.5, 142.1, 152.7, 166.6; IR (KBr) 677, 755, 865, 975, 1075, 1116, 1221, 1261, 1339, 1446, 1566, 1609, 2899, 2981, 3086, 3137, 3179, 3473. HRMS calcd for C14H12ClN2S 275.041, found 275.0412.

N-(4-Bromobenzyl)benzo[d]thiazol-2-amine (3g):[33] yield 94% (300 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.56 (d, J = 5.5 Hz, 2H), 7.02 (t, J = 7.5 Hz, 1H), 7.21 (t, J = 7.5 Hz, 1H), 7.33 (d, J = 8 Hz, 2H), 7.37 (d, J = 8 Hz, 1H), 7.53 (d, J = 8 Hz, 2H), 7.66 (d, J = 7.5 Hz, 1H), 8.53 (t, J = 5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 48.7, 120.5, 122.3, 123.3, 123.3, 127.8, 131.8, 132.7, 133.5, 140.8, 154.6, 168.4.

N-Benzylbenzo[d]thiazol-2-amine (3h):[33] yield 96% (230 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.60 (d, J = 5.5 Hz, 2H), 7.02 (t, J = 7.5 Hz, 1H), 7.20−7.27 (m, 2H), 7.33−7.40 (m, 5H), 7.66 (d, J = 8 Hz, 1H), 8.52 (t, J = 5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 47.4, 118.4, 121.3, 125.8, 127.3, 127.6, 128.6, 130.6, 139.2, 152.7, 166.5.

N-(Pyridin-3-ylmethyl)benzo[d]thiazol-2-amine (3i): yield 93% (224 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.62 (d, J = 5.5 Hz, 2H), 7.03 (t, J = 7.5 Hz, 1H), 7.23 (t, J = 7.5 Hz, 1H), 7.36−7.41 (m, 2H), 7.67 (d, J = 7.5 Hz, 1H), 7.78 (d, J = 8 Hz, 1H), 8.47 (d, J = 4 Hz, 1H), 8.56 (t, J = 5.5 Hz, 1H), 8.61 (s, 1H); ^{13}C NMR (125 MHz, DMSO) 45.2, 118.7, 121.4, 121.6, 123.9, 126.0, 130.9, 134.9, 135.7, 148.7, 149.3, 152.7, 166.5; IR (KBr) 629, 705, 762, 1049, 1101, 1122, 1208, 1433, 1538, 2362, 2921, 3167, 3428. HRMS calcd for C13H12N3S 242.0753, found 242.0774.

4-Methyl-N-(4-methoxybenzyl)benzo[d]thiazol-2-amine (3j): yield 91% (258 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.44 (s, 3H), 3.72 (s, 3H), 4.48 (d, J = 6 Hz, 2H), 6.89−6.92 (m, 3H), 7.03 (d, J = 7.5 Hz, 1H), 7.32 (d, J = 8.5 Hz, 2H), 7.46(d, J = 8 Hz, 1H), 8.41 (t, J = 5.5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 18.8, 47.6, 55.7, 114.4, 119.0, 121.4, 126.9, 127.9, 129.7, 130.5, 131.5, 152.0, 159.1, 166.1; IR (KBr) 625, 702, 755, 810, 833, 869, 888, 925, 963, 1027, 1069, 1174, 1215, 1245, 1285, 1353, 1404, 1449, 1513, 1534, 1594, 1762, 1828, 2362, 2626, 2839, 2914, 3231, 3866; HRMS calcd for C16H17N2OS 285.1062, found 285.1031.

N-Benzyl-4-methylbenzo[d]thiazol-2-amine (3k): yield 87% (221 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.43 (s, 3H), 4.57 (d, J = 5.5 Hz, 2H), 6.91 (t, J = 8 Hz, 1H), 7.03 (d, J = 7.5 Hz, 1H), 7.26 (t, J = 7 Hz, 1H), 7.34 (t, J = 8 Hz, 2H), 7.40 (d, J = 7 Hz, 2H), 7.46 (d, J = 7.5 Hz, 1H), 8.48 (t, J = 5.5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 18.0, 47.3, 118.2, 120.7, 126.1, 126.9, 127.1, 127.5, 128.2, 129.7, 138.8, 151.2, 165.4; IR (KBr) 617, 689, 742, 771, 881, 1027, 1077, 1209, 1242, 1453, 1523, 1556, 1644, 1733, 2359, 2921, 3209, 3437, 3645, 3732, 3796, 3922; HRMS calcd for C15H15N2S 255.0956, found 255.0948.

N-Benzyl-6-methylbenzo[d]thiazol-2-amine (3l):[33] yield 90% (228 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.30 (s, 3H), 4.58 (d, J = 5.5 Hz, 2H), 7.01 (d, J = 8 Hz, 1H), 7.24−7.28 (m, 1H), 7.32−7.38 (m, 4H), 7.45 (s, 1H), 8.39 (t, J = 5.5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 21.2, 47.6, 118.2, 121.3, 127.0, 127.4, 127.8, 128.8, 130.5, 130.9, 139.4, 150.7, 166.0.

N-(4-Chlorobenzyl)-6-methylbenzo[d]thiazol-2-amine (3m): yield 83% (239 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.31 (s, 3H), 4.56 (d, J = 5 Hz, 2H), 7.02 (d, J = 8 Hz, 1H), 7.26 (d, J = 8.5 Hz, 1H), 7.39 (s, 4H), 7.46 (s, 1H), 8.42 (t, J = 5 Hz, 1H) ;^{13}C NMR (125 MHz, DMSO) 21.2, 46.8, 118.3, 121.3, 127.0, 128.7, 129.6, 130.6, 130.9, 132.0, 138.6, 150.6, 165.9 ; IR (KBr) 677, 729, 814, 1014, 1086, 1262, 1348, 1464, 1616, 2364, 2854, 2912, 3452; MS m/z 289 [M + H]+, 243, 242, 214, 186, 125, 128, 104. Anal. Calcd for C15H13ClN2S: C, 62.38; H, 4.54; N, 9.70; S, 11.1. Found: C, 62.41; H, 4.52; N, 9.72; S, 11.07.

6-Chloro-N-(4-chlorobenzyl)benzo[d]thiazol-2-amine (3n): yield 90% (278 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.58 (d, J = 6 Hz, 2H), 7.23 (dd, J1 = 6.5 Hz, J2 = 2 Hz, 1H), 7.35−7.42 (m, 5H), 7.80 (d, J = 2 Hz, 1H), 8.67 (t, J = 6 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 46.9, 119.4, 121.1, 125.2, 126.1, 128.8, 129.6, 132.1, 132.5, 138.2, 151.7, 167.2; IR (KBr) 682, 726, 767, 815, 1090, 1257, 1345, 1445, 1570, 1613, 2360, 2903, 3085, 3452; HRMS calcd for C14H11Cl2N2S 309.0021, found 309.0016.

6-Chloro-N-(4-methoxybenzyl)benzo[d]thiazol-1-amine (3o): yield 82% (249 mg); ^{1}H NMR (200 MHz, DMSO) δ 3.72 (s, 3H), 4.49 (d, J = 4.6 Hz, 2H), 6.88 (d, J = 8.2 Hz, 2H), 7.19−7.38 (m, 4H), 7.77 (s, 1H), 8.55 (s, 1H); ^{13}C NMR (50 MHz, DMSO) 47.2, 55.4, 114.2, 119.3, 121.0, 125.1, 126.1, 129.3, 130.9, 132.5, 151.8, 158.9, 167.2; IR (KBr) 607, 767, 821, 1031, 1172, 1250, 1345, 1446, 1509, 1547, 1567, 1603, 2842, 2971, 3174, 3411; HRMS calcd for C15H14ClN2OS 305.0516, found 305.0517.

N-Benzyl-6-chlorobenzo[d]thiazol-2-amine (3p):[33] yield 89% (244 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.59 (d, J = 6 Hz, 2H), 7.22 − 7.28 (m, 2H), 7.33−7.38 (m, 5H), 7.79 (d, J = 2.5 Hz, 1H),

8.62 (t, *J* = 6 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 47.7, 119.4, 121.1, 125.1, 126.1, 127.5, 127.8, 128.8, 132.5, 139.1, 151.8, 167.3.

N-Benzyl-4,5-diphenylthiazol-2-amine (3q): yield 85% (290 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.49 (d, *J* = 5.5 Hz, 2H), 7.19 (d, *J* = 7 Hz, 2H), 7.23−7.30 (m, 7H), 7.34−7.41 (m, 6H), 8.26 (t, *J* = 6 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 47.4, 118.7, 126.9, 127.0 127.2, 127.4, 127.9, 128.2, 128.4, 128.6, 128.8, 132.5, 135.3, 139.0, 144.8, 165.8; IR (KBr) 690, 760, 971, 1068, 1230, 1324, 1417, 1580, 1956, 2356, 2962, 3085, 3195, 3416; HRMS calcd for $C_{22}H_{19}N_2S$ 343.1269, found 343.1266.

N-(4-Chlorobenzyl)-4,5-diphenylthiazol-2-amine (3r): yield 80% (300 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.48 (d, *J* = 6 Hz, 2H), 7.19 (d, *J* = 7 Hz, 2H), 7.24−7.30 (m, 6H), 7.38 (d, *J* = 6.5 Hz, 2H), 7.4 (s, 4H), 8.28 (t, *J* = 6 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 47.3, 119.7, 127.8, 128.0, 128.7, 128.9, 129.1, 129.4, 129.6, 130.0, 132.2, 133.2, 136.0, 138.9, 145.6, 166.4. IR (KBr) 694, 759, 1082, 1218, 1318, 1420, 1485, 1563, 1770, 1894, 1960, 2349, 2893, 2959, 3075, 3197, 3418; MS *m/z* 399 [M + Na]+, 377 [M + H]+, 375, 331, 302, 301, 261, 223, 217, 156, 140, 97, 81. Anal. Calcd for $C_{22}H_{17}ClN_2S$: C, 70.11; H, 4.55; N, 7.43; S, 8.51. Found: C, 70.17; H, 4.55; N, 7.40; S, 8.48.

N-Benzylpyrimidin-2-amine (3s):[17] yield 91% (168 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.49 (d, *J* = 6 Hz, 2H), 6.56 (t, *J* = 5 Hz, 1H) 7.18−7.21 (m, 1H), 7.27−7.29 (m, 4H), 7.70 (t, *J* = 5 Hz, 1H), 8.25 (d, *J* = 5 Hz, 2H); ^{13}C NMR (125 MHz, DMSO) 44.5, 110.8, 127.1, 127.6, 128.7, 141.07, 158.6, 162.9.

N-(4-Methylbenzyl)pyrimidin-2-amine (3t):[17] yield 90% (179 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.25 (s, 3H), 4.43 (d, *J* = 6.5 Hz, 2H), 6.55 (t, *J* = 4.5 Hz, 1H) 7.08 (d, *J* = 7.5 Hz, 2H), 7.17 (d, *J* = 8 Hz, 2H), 7.62 (t, *J* = 6.5 Hz, 1H), 8.24 (d, *J* = 5 Hz, 2H); ^{13}C NMR (125 MHz, DMSO) 21.3, 44.2, 110.8, 127.6, 129.3, 136.1, 138.0, 158.6, 162.9.

N-(4-Chlorobenzyl)pyrimidin-2-amine (3u):[17] yield 93% (204 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.45 (d, *J* = 6.5 Hz, 2H), 6.57 (t, *J* = 5 Hz, 1H) 7.30−7.35 (m, 4H), 7.72 (t, *J* = 6.5 Hz, 1H), 8.25 (d, *J* = 5 Hz, 2H); ^{13}C NMR (125 MHz, DMSO) 43.9, 111.0, 128.7, 129.5, 131.6, 140.2, 158.6, 162.8.

N-Benzylpyridin-2-amine (3v):[17] yield 90% (165 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.46 (d, *J* = 6 Hz, 2H), 6.46 (t, *J* = 6 Hz, 1H), 6.49 (d, *J* = 8.5 Hz, 1H) 7.01 (t, *J* = 5.5 Hz, 1H), 7.20 (t, *J* = 7 Hz, 1H), 7.27−7.36 (m, 5H), 7.94 (d, *J* = 5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 44.8, 108.8, 112.3, 127.1, 127.8, 128.8, 137.3, 141.3, 148.2, 159.3.

N-(4-Methylbenzyl)pyridin-2-amine (3w):[17] yield 92% (182 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.25 (s, 3H), 4.40 (d, *J* = 6 Hz, 2H), 6.44−6.48 (m, 2H), 6.94 (s, 1H), 7.09 (d, *J* = 7 Hz, 2H), 7.19 (d, *J* = 7.5 Hz, 2H), 7.33 (t, *J* = 7.5 Hz, 1H), 7.93 (d, *J* = 3.5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 21.3, 44.5, 108.7, 112.3, 127.8, 129.4, 136.1, 137.2, 138.2, 148.2, 159.3.

N-Benzylpyrazin-2-amine (3x): yield 92% (170 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.47 (d, *J* = 6 Hz, 2H), 7.21−7.25 (m, 1H), 7.30−7.34 (m, 4H), 7.58 (t, *J* = 5.5 Hz, 1H), 7.65 (d, *J* = 3 Hz, 1H), 7.91 (s, 1H), 7.97 (s, 1H); ^{13}C NMR (125 MHz, DMSO) 44.2, 127.4, 127.9, 128.9, 131.9, 133.9, 140.4, 142.2, 155.5; IR (KBr) 696, 753, 822, 1001, 1056, 1324, 1426, 1567, 1771, 1897, 2356, 3039, 3088, 3231, 3425; HRMS calcd for $C_{11}H_{12}N_3$ 186.1031, found 186.1010.

N-(4-Methylbenzyl)pyrazin-2-amine (3y): yield 90% (179 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.26 (s, 3H), 4.41 (d, *J* = 6 Hz, 2H), 7.11 (d, *J* = 7.5, Hz, 2H), 7.20 (d, *J* = 8 Hz, 2H), 7.51 (d, *J* = 5 Hz, 1H), 7.64 (d, *J* = 2.5 Hz, 1H), 7.90 (s, 1H), 7.95 (s, 1H); ^{13}C NMR (125 MHz, DMSO) 21.3, 44.0, 127.9, 129.5, 131.85, 133.9, 136.4, 137.3, 142.1, 155.5; IR (KBr) 686, 807, 1002, 1050, 1103, 1144, 1198, 1329, 1429, 1571, 1910, 2358, 2996, 3094, 3231, 3427; HRMS calcd for $C_{12}H_{14}N_3$ 200.1188, found 200.1176.

N-(4-Chlorobenzyl)pyrazin-2-amine (3z): yield 91% (199 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.46 (d, *J* = 6 Hz, 2H), 7.33−7.38 (m, 4H), 7.62 (t, *J* = 6 Hz, 1H) 7.66 (d, *J* = 2.5 Hz, 1H), 7.90 (s, 1H), 7.97 (s, 1H); ^{13}C NMR (125 MHz, DMSO) 43.5, 128.8, 129.7, 132.0, 133.9, 139.5, 142.1, 155.4; IR (KBr) 695, 799, 1002, 1090, 1193, 1319, 1428, 1577, 1900, 2354, 2930, 3243, 3424; HRMS calcd for $C_{12}H_{14}N_3$ 220.0642, found 220.0672.

Synthesis of *N*,5-Dibenzyl-4-phenylthiazol-2-amine (5a). In a 25 mL round bottomed flask were placed 176 mg (1.0 mmol) of 4-phenylthiazol-2-amine **4a**, 324 mg (3.0 mmol) of benzyl alcohol, 40 mg (1.0 mmol) of sodium hydroxide (NaOH) and 1.0 mL of toluene. The reaction flask was heated at130 °C for 28 h in an oil bath. After completion of the reaction, the flask was allowed to attain room temperature, and 20 mL of water was added. The product was extracted with EtOAc (30 mL × 3). After removal of the organic solvent under reduced pressure, the residue was subjected to column chromatography on silica gel using dichloromethane/ethyl acetate (95:5) as eluent. After removal of the solvent, **5a** was obtained in 81% yield (288 mg; 0.81 mmol): ^{1}H NMR (500 MHz, DMSO) ppm 4.08 (s, 2H), 4.45 (d, *J* = 5.5 Hz, 2H), 7.19−7.26 (m, 4H), 7.28−7.34 (m, 5H), 7.39 (t, *J* = 7.5 Hz, 4H), 7.57 (d, *J* = 7.5 Hz, 2H), 8.00 (t, *J* = 5.5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 32.6, 47.9, 119.3, 126.8, 127.4, 127.6, 127.9, 128.4, 128.7, 129.0, 135.8, 139.7, 140.9, 146.5, 166.0; IR (KBr) 702, 845, 1025, 1165, 1232, 1331, 1423, 1449, 1490, 1569, 2949, 3024, 3183, 3450; HRMS calcd for $C_{23}H_{21}N_2S$ 357.1425, found 357.1455.

N-Benzyl-4-phenylthiazol-2-amine (5a'): yield 12% (32 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.51 (d, *J* = 6 Hz, 2H), 7.04 (s, 1H) 7.25 (t, *J* = 7.5 Hz, 2H), 7.32−7.41 (m, 6H), 7.82 (d, *J* = 7.5 Hz, 2H), 8.18 (t, *J* = 5.5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 48.2, 101.6, 126.0, 127.4, 127.7, 128.0, 128.7, 128.9, 135.2, 139.7, 150.3, 168.8; IR (KBr) 698, 726, 974, 1023, 1065, 1106, 1334, 1424, 1482, 1585, 2892, 2929, 2977, 3025, 3057, 1101, 3212, 3429; HRMS calcd for $C_{16}H_{15}N_2S$ 267.0957, found 267.0973.

N,5-Bis(4-methoxybenzyl)-4-phenylthiazol-2-amine (5b): yield 78% (324 mg); ^{1}H NMR (500 MHz, DMSO) δ 3.70 (s, 3H), 3.71 (s, 3H), 4.00 (s, 2H), 4.35 (d, *J* = 6 Hz, 2H), 6.85−6.90 (m, 4H), 7.09 (d, *J* = 8.5 Hz, 2H), 7.28−7.31 (m, 3H), 7.38 (t, *J* = 7.5 Hz, 2H), 7.57 (d, *J* = 7 Hz, 2H) 7.89 (t, *J* = 6 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 31.8, 47.4, 55.4, 114.1, 114.3, 120.0, 127.6, 128.4, 128.7, 129.3, 129.5, 131.6, 132.8, 135.8, 146.1, 158.2, 158.7, 165.9; IR (KBr) 695, 804, 1030, 1164, 1245, 1294, 1427, 1510, 1590, 2898, 2939, 3098, 3204, 3425; HRMS calcd for $C_{25}H_{25}N_2O_2S$ 417.1637, found 417.1653.

N,5-Bis(4-chlorobenzyl)-4-phenylthiazol-2-amine (5c): yield 82% (348 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.07 (s, 2H), 4.42 (d, *J* = 6 Hz, 2H), 7.20 (d, *J* = 8.5 Hz, 2H), 7.31 (t, *J* = 7.5 Hz, 1H), 7.35−7.41 (m, 8H), 7.51 (d, *J* = 7.5 Hz, 2H), 8.05 (t, *J* = 6 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 31.4, 46.6, 118.3, 127.2, 127.9, 128.1, 128.4, 129.2, 129.8, 130.9, 130.9, 131.4, 135.1, 138.3, 139.3, 146.2, 165.4; IR (KBr) 700, 774, 808, 1012, 1088, 1238, 1326, 1408, 1487, 1571, 1738, 2924, 2956, 3081, 3188, 3436; HRMS calcd for $C_{23}H_{19}Cl_2 N_2S$ 425.0646, found 425.0657.

N,5-Dibenzyl-4-(4-tert-butylphenyl)thiazol-2-amine (5d): yield 83% (342 mg); ^{1}H NMR (500 MHz, DMSO) δ 1.28 (s, 9H), 4.07 (s, 2H), 4.42 (d, *J* = 6 Hz, 2H), 7.19−7.25 (m, 4H), 7.29−7.40 (m, 8H), 7.47 (d, *J* = 8.5 Hz, 2H), 7.96 (t, *J* = 6 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 31.5, 32.6, 34.7, 47.8, 118.7, 125.4, 126.8, 127.3, 127.9, 128.1, 128.4, 128.7, 129.0, 133.0, 139.8, 140.9, 146.4, 150.0, 165.9; IR (KBr) 711, 844, 973, 1023, 1111, 1163, 1329, 1422, 1451, 1494, 1593, 2363, 2898, 2959, 3105, 3220, 3426; HRMS calcd for $C_{27}H_{29}N_2S$ 413.2051, found 413.2071.

4-(4-tert-Butylphenyl)-N,5-bis(4-methylbenzyl)thiazol-2-amine (5e): yield 80% (352 mg); ^{1}H NMR (500 MHz, DMSO) δ 1.28 (s, 9H), 2.26 (s, 6H), 4.00 (s, 2H), 4.35 (d, *J* = 5 Hz, 2H), 7.06−7.13 (m, 6H), 7.23 (d, *J* = 7.5 Hz, 2H), 7.38 (d, *J* = 8 Hz, 2H), 7.46 (d, *J* = 8 Hz, 2H), 7.86 (t, *J* = 5.5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 20.7, 20.8, 31.2, 32.0, 34.4, 47.4, 118.7, 125.1, 127.6, 127.8, 128.0, 128.9, 129.2, 132.7, 135.5, 136.1, 136.4, 137.6, 146.0, 149.7, 165.5; IR (KBr) 678, 793, 842, 980, 1019, 1109, 1169, 1331, 1424, 1460, 1509, 1594, 1799, 1894, 2365, 2961, 3102, 3208, 3423; HRMS calcd for $C_{29}H_{33}N_2S$ 441.2364, found 441.2379.

4-(4-tert-Butylphenyl)-N,5-bis(4-methoxybenzyl)thiazol-2-amine (5f): yield 80% (377 mg); ^{1}H NMR (500 MHz, DMSO) δ 1.28 (s, 9H), 3.71 (s, 3H), 3.72 (s, 3H), 3.99 (s, 2H), 4.33 (d, *J* = 5.5 Hz, 2H), 6.85−6.89 (m, 4H), 7.10 (d, *J* = 7.5 Hz, 2H), 7.28 (d, *J* = 7.5 Hz, 2H), 7.39 (d, *J* = 7 Hz, 2H), 7.49 (d, *J* = 7.5 Hz, 2H), 7.85 (t, *J* = 5.5 Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 31.5, 31.8, 34.7, 47.4, 55.4, 114.1,

114.3, 119.4, 125.4, 128.1, 129.3, 129.4, 131.7, 132.9, 133.1, 146.1, 149.9, 158.2, 158.7, 165.8; IR (KBr) 678, 730, 808, 1031, 1106, 1168, 1248, 1427, 1459, 1510, 1593, 1876, 1919, 2361, 2959, 3212, 3423; HRMS calcd for $C_{29}H_{33}N_2O_2S$ 473.2262, found 473.2288.

4-(4-tert-Butylphenyl)-N,5-bis(4-chlorobenzyl)thiazol-2-amine (5g): yield 81% (389 mg); ^{1}H NMR (500 MHz, DMSO) δ 1.27 (s, 9H), 4.06 (s, 2H), 4.41 (d, $J = 6$ Hz, 2H), 7.20 (d, $J = 8.5$ Hz, 2H), 7.35−7.40 (m, 8H), 7.44 (d, $J = 8.5$ Hz, 2H), 8.02 (t, $J = 6$ Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 31.5, 31.9, 34.7, 47.1, 118.2, 125.5, 128.1, 128.6, 128.9, 129.8, 130.3, 131.4, 131.9, 132.8, 138.9, 139.9, 146.7, 150.1, 165.8; IR (KBr) 687, 729, 811, 1013, 1090, 1267, 1328, 1360, 1406, 1458, 1489, 1575, 1896, 2362, 2962, 3087, 3197, 3424; HRMS calcd for $C_{27}H_{27}Cl_2N_2S$ 481.1272, found 481.1278.

4-(4-tert-Butylphenyl)-N,5-bis(2-chlorobenzyl)thiazol-2-amine (5h): yield 85% (408 mg); ^{1}H NMR (500 MHz, DMSO) δ 1.27 (s, 9H), 4.15 (s, 2H), 4.52 (d, $J = 5.5$ Hz, 2H), 7.24−7.35 (m, 5H), 7.39 (d, $J = 8.5$ Hz, 2H), 7.43−7.49 (m, 5H), 8.01 (t, $J = 5.5$ Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 30.5, 31.2, 34.4, 45.1, 116.9, 125.2, 127.3, 127.7, 127.7, 128.7, 128.9, 129.3, 129.5, 129.5, 130.3, 132.5, 132.5, 132.9, 136.4, 137.9, 146.4, 149.9, 165.2; IR (KBr) 691, 750, 841, 1040, 1118, 1168, 1242, 1326, 1442, 1587, 2874, 2961, 3094, 3200, 3425; HRMS calcd for $C_{27}H_{27}Cl2N_2S$ 481.1273, found 481.1301.

N,5-Dibenzyl-4-p-tolylthiazol-2-amine (5i): yield 80% (296 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.30 (s, 3H), 4.05 (s, 2H), 4.42 (d, $J = 5.5$ Hz, 2H), 7.17−7.25 (m, 6H), 7.28−7.37 (m, 6H), 7.44 (d, $J = 8$ Hz, 2H), 7.96 (t, $J = 6$ Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 20.9, 32.3, 47.6, 118.4, 126.5, 127.0, 127.6, 128.0, 128.1, 128.4, 128.7, 129, 13.76, 136.6, 139.4, 140.7, 146.2, 165.6; IR (KBr) 693, 723, 827, 975, 1163, 1239, 1329, 1425, 1447, 1495, 1592, 2897, 3205, 3433; HRMS calcd for $C_{24}H_{23}N_2S$ 371.1583, found 371.1590.

N,5-Bis(2-chlorobenzyl)-4-p-tolylthiazol-2-amine (5j): yield 81% (355 mg); ^{1}H NMR (500 MHz, DMSO) δ 2.30 (s, 3H), 4.14 (s, 2H), 4.51 (d, $J = 5.5$ Hz, 2H), 7.18 (d, $J = 8$ Hz, 2H), 7.24−7.35 (m, 5H), 7.39 (d, $J = 8$ Hz, 2H), 7.43 (d, $J = 7.5$ Hz, 2H), 7.47 (d, $J = 7.5$ Hz, 1H), 7.99 (t, $J = 6$ Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 20.7, 30.3, 45.0, 116.7, 127.0, 127.4, 127.7, 128.4, 128.7, 128.8, 129.1, 129.3, 130.1, 132.3, 132.4, 132.7, 136.2, 136.4, 137.8, 146.3, 165.1; IR (KBr) 695, 744, 825, 1040, 1168, 1234, 1326, 1443, 1505, 1589, 2362, 2890, 2927, 2963, 3064, 3197, 3429; HRMS calcd for $C_{24}H_{21}Cl_2N_2S$ 439.0803, found 439.0813.

N,5-Dibenzyl-4-(4-chlorophenyl)thiazol-2-amine (5k): yield 80% (312 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.07 (s, 2H), 4.42 (d, $J = 6$ Hz, 2H), 7.18−7.26 (m, 4H), 7.29−7.36 (m, 6H), 7.44 (d, $J = 8.5$ Hz, 2H), 7.57 (d, $J = 8.5$ Hz, 2H), 8.01 (t, $J = 6$ Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 32.6, 47.8, 120.1, 126.9, 127.4, 127.9, 128.5, 128.7, 129.0, 130.1, 132.1, 134.6, 139.6, 140.6, 145.0, 166.0; IR (KBr) 696, 754, 835, 973, 1012, 1087, 1164, 1426, 1448, 1483, 1591, 2360, 2898, 3207, 3423; HRMS calcd for $C_{23}H_{20}ClN_2S$ 391.1036, found 391.1041.

N,5-Bis(2-chlorobenzyl)-4-(4-chlorophenyl)thiazol-2-amine (5l): yield 82% (376 mg); ^{1}H NMR (500 MHz, DMSO) δ 4.16 (s, 2H), 4.53 (d, $J = 6$ Hz, 2H), 7.26−7.34 (m, 5H), 7.43−7.49 (m, 5H), 7.55 (d, $J = 8.5$ Hz, 2H), 8.06 (t, $J = 5.5$ Hz, 1H); ^{13}C NMR (125 MHz, DMSO) 30.5, 45.2, 118.5, 127.3, 127.7, 128.5, 128.8, 128.9, 129.3, 129.5, 129.7, 130.4, 131.9, 132.5, 132.9, 134.2, 136.3, 137.7, 145.0, 165.45; IR (KBr) 693, 746, 832, 1040, 1089, 1164, 1234, 1328, 1443, 1587, 2891, 3197, 3419; HRMS calcd for $C_{23}H_{18}Cl3N_2S$ 459.0256, found 459.0258.

ASSOCIATED CONTENT

Ⓢ Supporting Information

Copies of NMR spectra for all compounds and HRMS spectra for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org. Crystallographic data for compound **5h** (CCDC- 916195) can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

AUTHOR INFORMATION

Corresponding Author

*E-mail: adimurthy@csmcri.org.

Notes

The authors declare no competing financial interest.
‡Dissertation student at CSIR-CSMCRI.

REFERENCES

(1) Smith, M. B.; March, J. In *March's Advanced Organic Chemistry*, 6th ed.; Wiley-Interscience: New York, 2007.

(2) Baxter, E. W.; Reitz, A. B. *Organic Reactions*; John Wiley & Sons: New York, 2004.

(3) Hartwig, J. F. H. *Acc. Chem. Res.* **2008**, *41*, 1534.

(4) Surry, D. A.; Buchwald, S. L. *Angew. Chem., Int. Ed.* **2008**, *47*, 6338.

(5) Lawerencem, S. A. *Amines: Synthesis Properties and Applications*; Cambridge University: Cambridge, 2004.

(6) Vollhardt, K. P. C.; Schore, N. E. *Organic Chemistry: Structure and Function*, 5th ed., W H. Freeman: New York, 1999.

(7) Nunez Magro, A. A.; Eastham, G. R.; Cole-Hamilton, D. J. *Chem. Commun.* **2007**, 3154.

(8) Seayad, A.; Ahmed, M.; Klein, H.; Jackstell, R.; Gross, T.; Beller, M. *Science* **2002**, *297*, 1676.

(9) Li, F.; Kang, Q.; Shan, H.; Chen, L.; Xie, J. *Eur. J. Org. Chem.* **2012**, 5085 and references therein.

(10) Guillena, G.; Ramsón, D. J.; Yus, M. *Chem. Rev.* **2010**, *110*, 1611.

(11) Nef, J. U. *Justus Liebigs Ann. Chem.* **1901**, *318*, 137.

(12) Lazier, W. A.; Adkins, H. *J. Am. Chem. Soc.* **1924**, *46*, 741.

(13) Sprinzak, Y. *J. Am. Chem. Soc.* **1956**, *78*, 3207.

(14) Miyano, S. *Chem. Pharm. Bull.* **1965**, *13*, 1135.

(15) Zvezdina, A.; Pozharskii, A. F.; Sokolov, V. I. *Chem. Heterocycl. Compd.* **1970**, *6*, 389.

(16) Miyano, S.; Nakao, M. *Chem. Pharm. Bull.* **1972**, *20*, 1328.

(17) Xu, Q.; Li, Q.; Zhu, X.; Chen, J. *Adv. Synth. Catal.* **2013**, *355*, 73.

(18) Donthiri, R. R.; Patil, R. D.; Adimurthy, S. *Eur. J. Org. Chem.* **2012**, 4457.

(19) Tang and co-workers have also described the activation of alcohol by base in presence of amines; see: Xu, J.; Zhuang, R.; Bao, L.; Tang, G.; Zhao, Y. *Green Chem.* **2012**, *14*, 2384.

(20) Mohan, D. C.; Rao, S. N.; Adimurthy, S. *J. Org. Chem.* **2013**, *78*, 1266.

(21) Patil, R. D.; Adimurthy, S. *Adv. Synth. Catal.* **2011**, *353*, 1695.

(22) Chunavala, K. C.; Joshi, G.; Suresh, E.; Adimurthy, S. *Synthesis* **2011**, 635.

(23) Chunavala, K. C.; Adimurthy, S. *Synth. Commun.* **2011**, *41*, 1843.

(24) Rao, S. N.; Mohan, D. C.; Adimurthy, S. *Org. Lett.* **2013**, *15*, 1496.

(25) Cavalleri, B.; Turconi, M.; Tamborini, G.; Occelli, E.; Cietto, G.; Pallanza, R.; Scotti, R.; Berti, M.; Roman, G.; Parenti, F. *J. Med. Chem.* **1990**, *33*, 1470.

(26) Das, J.; Moquin, R. V.; Lin, J.; Liu, C.; Doweyko, A. M.; Defex, H. F.; Fang, Q.; Pang, S.; Pitt, S.; Shen, D. R.; Schieven, G. L.; Barrish, J. C.; Wityak, J. *Bioorg. Med. Chem. Lett.* **2003**, *13*, 2587.

(27) Byeon, S. R.; Jin, Y. J.; Lim, S. J.; Lee, J. H.; Yoo, K. H.; Jung, S. K.; Ohb, S. J.; Kim, D. J. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 4022.

(28) Sun, Z. Y.; Zhu, Z.; Ye, Y.; McKittrick, B.; Czarniecki, M.; Greenlee, W.; Mullins, D.; Guzzi, M. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 6801.

(29) Basavarajaiah, S. M.; Mruthyujayaswamy, B. H. M. *Indian J. Chem.* **2010**, *49B*, 1117.

(30) El-Subbagh, H. I.; Abadi, A. H.; Lehmann, J. *Arch. Pharm. Pharm. Med. Chem.* **1999**, *332*, 137.

(31) Dykhuizen, E. C.; May, J. F.; Tongpenyai, A.; Kiessling, L. L. *J. Am. Chem. Soc.* **2008**, *130*, 6706.

(32) Zhang, T.; Yan, Z.; Sromek, A.; Knapp, B. I.; Scrimale, T.; Bidlack, J. M.; Neumeyer, J. L. *J. Med. Chem.* **2011**, *54*, 1903.

(33) Li, F.; Shan, H.; Kang, Q.; Chen, L. *Chem. Commun.* **2011**, *47*, 5058.

(34) Li, F.; Shan, H.; Chen, L.; Kang, Q.; Zou, P. *Chem. Commun.* **2012**, *48*, 603.

(35) Zhang, W.; Yue, Y.; Yu, D.; Song, L.; Xu, Y. Y.; Tian, Y. J .; Guo, Y. J. *Adv. Synth. Catal.* **2012**, *354*, 2283.

(36) Martinez-Asencio, A.; Yus, M.; Ramon, D. J. *Synthesis* **2011**, 3730.

(37) Ma, D.; Lu, X.; Shi, L.; Zhang, H.; Jiang, Y.; Liu, X. *Angew. Chem., Int. Ed.* **2011**, *50*, 1118.

(38) Saha, P.; Ramana, T.; Purkait, N.; Ali, M. A.; Paul, R.; Punniyamurthy, T. *J. Org. Chem.* **2009**, *74*, 8719.

(39) Cho, S. H.; Kim, J. Y.; Lee, S. Y.; Chang, S. *Angew. Chem., Int. Ed.* **2009**, *48*, 9127.

(40) Wang, X.; Zhang, B.; Wang, D. Z. *Tetrahedron* **2011**, *67*, 3406.

(41) Lewis, G. E. *J. Org. Chem.* **1965**, *30*, 2433.